# Chemical Book India

| <b>Chemical Safety</b> | <b>Data Sheet</b> | MSDS / SDS | 5 |
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# Phosphoryl tribromide SDS

Revision Date: 2024-04-25 Revision Number: 1

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# SECTION 1: Identification of the substance/mixture and of the company/undertaking

| Product identifier |                       |
|--------------------|-----------------------|
| Product name:      | Phosphoryl tribromide |
| CAS:               | 7789-59-5             |

## Relevant identified uses of the substance or mixture and uses advised against

 Relevant identified
 For R&D use only. Not for medicinal, household or other use.

 uses:
 uses advised

 uses:
 none

 against:

## **Company Identification**

| Company:   | Chemicalbook.in   |
|------------|---|
| Address:   | 5 vasavi Layout Basaveswara Nilayam Pragathi Nagar Hyderabad, India -500090 |
| Telephone: | +91 9550333722  |

# **SECTION 2: Hazards identification**

Classification of the substance or mixture

Skin corrosion, Sub-category 1B

### GHS label elements, including precautionary statements

Pictogram(s)

Signal word Danger

# Hazard statement(s)

H314 Causes severe skin burns and eye damage

### Precautionary statement(s)

### Prevention

P260 Do not breathe dust/fume/gas/mist/vapours/spray. P264 Wash ... thoroughly after handling. P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

#### Response

P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
P363 Wash contaminated clothing before reuse.
P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P316 Get emergency medical help immediately.
P321 Specific treatment (see ... on this label).
P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

### Storage

P405 Store locked up.

### Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

### Other hazards which do not result in classification

no data available

# SECTION 3: Composition/information on ingredients

| Substance                  |                       |
|----------------------------|-----------------------|
| Chemical name:             | Phosphoryl tribromide |
| Common names and synonyms: | Phosphoryl tribromide |
| CAS number:                | 7789-59-5             |
| EC number:                 | 232-177-7             |
| Concentration:             | 100%                  |

# **SECTION 4: First aid measures**

## Description of necessary first-aid measures

## lf inhaled

Move the victim into fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration and consult a doctor immediately. Do not use mouth to mouth resuscitation if the victim ingested or inhaled the chemical.

## Following skin contact

Take off contaminated clothing immediately. Wash off with soap and plenty of water. Consult a doctor.

## Following eye contact

Rinse with pure water for at least 15 minutes. Consult a doctor.

## Following ingestion

Rinse mouth with water. Do not induce vomiting. Never give anything by mouth to an unconscious person. Call a doctor or Poison Control Center immediately.

## Most important symptoms/effects, acute and delayed

Excerpt from ERG Guide 137 [Substances - Water-Reactive - Corrosive]: CORROSIVE and/or TOXIC; inhalation, ingestion or contact (skin, eyes) with vapors, dusts or substance may cause severe injury, burns or death. Fire will produce irritating, corrosive and/or toxic gases. Reaction with water may generate much heat that will increase the concentration of fumes in the air. Contact with molten substance may cause severe burns to skin and eyes. Runoff from fire control or dilution water may cause pollution. (ERG, 2016)

### Indication of immediate medical attention and special treatment needed, if necessary

no data available

# **SECTION 5: Firefighting measures**

### Suitable extinguishing media

Excerpt from ERG Guide 137 [Substances - Water-Reactive - Corrosive]: When material is not involved in fire, do not use water on material itself. SMALL FIRE: Dry chemical or CO2. Move containers from fire area if you can do it without risk. LARGE FIRE: Flood fire area with large quantities of water, while knocking down vapors with water fog. If insufficient water supply: knock down vapors only. FIRE INVOLVING TANKS OR CAR/TRAILER LOADS: Cool containers with flooding quantities of water until well after fire is out. Do not get water inside containers. Withdraw immediately in case of rising sound from venting safety devices or discoloration of tank. ALWAYS stay away from tanks engulfed in fire. (ERG, 2016)

### Specific hazards arising from the chemical

Excerpt from ERG Guide 137 [Substances - Water-Reactive - Corrosive]: EXCEPT FOR ACETIC ANHYDRIDE (UN1715), THAT IS FLAWWABLE, some of these materials may burn, but none ignite readily. May ignite combustibles (wood, paper, oil, clothing, etc.). Substance will react with water (some violently), releasing corrosive and/or toxic gases and runoff. Flammable/toxic gases may accumulate in confined areas (basement, tanks, hopper/tank cars, etc.). Contact with metals may evolve flammable hydrogen gas. Containers may explode when heated or if contaminated with water. Substance may be transported in a molten form. (ERG, 2016)

### Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

# **SECTION 6: Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

### Environmental precautions

Prevent further spillage or leakage if it is safe to do so. Do not let the chemical enter drains. Discharge into the environment must be avoided.

### Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

# SECTION 7: Handling and storage

#### Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### Conditions for safe storage, including any incompatibilities

Should be stored in a cool, well-ventilated place, out of direct rays of sun, away from areas of high fire hazard, &...periodically inspected & monitored. incompatible materials should be isolated...

# SECTION 8: Exposure controls/personal protection

Control parameters

## Occupational Exposure limit values

no data available

#### Biological limit values

no data available

### Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

# Individual protection measures, such as personal protective equipment (PPE)

#### Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

# Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

# Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

# Thermal hazards

no data available

# SECTION 9: Physical and chemical properties and safety characteristics

| Physical state:   | Phosphorus oxybromide is a colorless crystalline solid or liquid if heated above 133° F with<br>a pungent odor. It is commonly heated and shipped in a molten state. Soluble in water, but,<br>decomposed by water to hydrobromic and phosphoric acid with evolution of heat. Reacts<br>with organic materials to cause fire. Evolves highly toxic and corrosive gases when<br>exposed to fire. Corrosive to metals and tissue. Used to make other chemicals. |
|---|---|
| Colour:   | COLORLESS CRYSTALS  |
| Odour:  | no data available   |
| Melting<br>point/freezing<br>point:                             | 56°C  |
| Boiling point or<br>initial boiling point<br>and boiling range: | 192°C(lit.)   |
| Flammability:   | no data available   |
| Lower and upper<br>explosion<br>limit/flammability<br>limit:    | no data available   |
| Flash point:  | 189°C   |
| Auto-ignition<br>temperature:                                   | no data available   |

| Decomposition<br>temperature:                 | no data available   |
|---|---|
| pH:   | no data available   |
| Kinematic<br>viscosity:                       | no data available   |
| Solubility:                                   | SOL IN ETHER, BENZENE, CHLOROFORM, CARBON DISULFIDE, CONCENTRATED SULFURIC ACID |
| Partition<br>coefficient n-<br>octanol/water: | no data available   |
| Vapour pressure:                              | no data available   |
| Density and/or relative density:              | 2.82g/mLat 25°C(lit.)   |
| Relative vapour<br>density:                   | no data available   |
| Particle<br>characteristics:                  | no data available   |

# SECTION 10: Stability and reactivity

## Reactivity

Reacts slowly and exothermically with water to form phosphoric acid and hydrobromic acid [Merck, 11th ed. 1989]. Decomposes exothermically in water to form toxic fumes of hydrobromic and phosphoric acids [Lewis]. Experiments conducted at Argonne showed that HBr gas is generated upon mixing with water. However, it was shown that nearly all of the generated HBr was apparently dissolved into the excess water as rapidly as it formed. So that virtually none of the gas was observed above the solution. "Development of the Table of Initial Isolation and Protective Distances for the 2008 Emergency Response Guidebook", ANL/DIS-09-2, D.F. Brown, H.M. Hartmann, W.A. Freeman, and W.D. Haney, Argonne National Laboratory, Argonne, Illinois, June 2009.

# Chemical stability

no data available

## Possibility of hazardous reactions

PHOSPHORUS OXYBROMIDE is incompatible with water, strong oxidizing agents, alcohols, bases, including amines. May react vigorously or explosively if mixed with disopropyl ether or other ethers in the presence of trace amounts of metal salts [J. Haz. Mat., 1981, 4, 291].

### Conditions to avoid

no data available

## Incompatible materials

Reacts strongly with organic matter

## Hazardous decomposition products

When heated to decomp, it emits highly toxic fumes of bromides & po(x).

# SECTION 11: Toxicological information

Acute toxicity Oral: no data available Inhalation: no data available Dermal: no data available

## Skin corrosion/irritation

no data available

## Serious eye damage/irritation

no data available

### Respiratory or skin sensitization

no data available

# Germ cell mutagenicity

no data available

Carcinogenicity

### Reproductive toxicity

no data available

STOT-single exposure

no data available

# STOT-repeated exposure

no data available

# Aspiration hazard

no data available

# SECTION 12: Ecological information

Toxicity Toxicity to fish: no data available Toxicity to daphnia and other aquatic invertebrates: no data available Toxicity to algae: no data available Toxicity to microorganisms: no data available

### Persistence and degradability

no data available

## Bioaccumulative potential

no data available

### Mobility in soil

no data available

### Other adverse effects

no data available

# **SECTION 13: Disposal considerations**

### Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

### **UN Number**

ADR/RID: UN2576 (For reference only, please check.) IMDG: UN2576 (For reference only, please check.) IATA: UN2576 (For reference only, please check.)

### **UN Proper Shipping Name**

ADR/RID: PHOSPHORUS OXYBROWIDE, MOLTEN (For reference only, please check.) IMDG: PHOSPHORUS OXYBROWIDE, MOLTEN (For reference only, please check.) IATA: PHOSPHORUS OXYBROWIDE, MOLTEN (For reference only, please check.)

## Transport hazard class(es)

ADR/RID: 8 (For reference only, please check.)

IMDG: 8 (For reference only, please check.) IATA: 8 (For reference only, please check.)

# Packing group, if applicable

ADR/RID: II (For reference only, please check.) IMDG: II (For reference only, please check.) IATA: II (For reference only, please check.)

## Environmental hazards

ADR/RID: No IMDG: No IATA: No

### Special precautions for user

no data available

## Transport in bulk according to IMO instruments

no data available

# SECTION 15: Regulatory information

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

# EC Inventory

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Listed.

New Zealand Inventory of Chemicals (NZIoC)

Listed.

(PICCS)

Listed.

Vietnam National Chemical Inventory

Not Listed.

IECSC)

Listed.

Korea Existing Chemicals List (KECL)

Listed.

# **SECTION 16: Other information**

### Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en

CAWEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

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