# Chemical Safety Data Sheet MSDS / SDS

## Perfluorohexane-1-sulphonic acid SDS

Revision Date: 2024-04-25 Revision Number: 1

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# SECTION 1: Identification of the substance/mixture and of the company/undertaking

#### Product identifier

Product name: Perfluorohexane-1-sulphonic acid

none

CAS: 355-46-4

### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified For R&D use only. Not for medicinal, household or other use.

uses:

Uses advised

against:

## Company Identification

Company: Chemicalbook.in

Address: 5 vasavi Layout Basaveswara Nilayam Pragathi Nagar Hyderabad, India -500090

Telephone: +91 9550333722

## **SECTION 2: Hazards identification**

### Classification of the substance or mixture

Acute toxicity - Category 4, Oral Acute toxicity - Category 4, Dermal Skin corrosion, Sub-category 1B Acute toxicity - Category 4, Inhalation

## GHS label elements, including precautionary statements

Pictogram(s)



Signal word Danger

## Hazard statement(s)

H302 Harmful if swallowed

H312 Harmful in contact with skin

H314 Causes severe skin burns and eye damage

H332 Harmful if inhaled

## Precautionary statement(s)

#### Prevention

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

P260 Do not breathe dust/fume/gas/mist/vapours/spray.

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P271 Use only outdoors or in a well-ventilated area.

## Response

P301+P317 IF SWALLOWED: Get medical help.

P330 Rinse mouth.

P302+P352 IF ON SKIN: Wash with plenty of water/...

P317 Get medical help.

P321 Specific treatment (see ... on this label).

P362+P364 Take off contaminated clothing and wash it before reuse.

P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P363 Wash contaminated clothing before reuse.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P316 Get emergency medical help immediately.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do.

Continue rinsing.

### Storage

P405 Store locked up.

## Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

#### Other hazards which do not result in classification

no data available

# **SECTION 3: Composition/information on ingredients**

#### Substance

Chemical name: Perfluorohexane-1-sulphonic acid
Common names and Perfluorohexane-1-sulphonic acid

synonyms:

CAS number: 355-46-4
EC number: 206-587-1
Concentration: 100%

### **SECTION 4: First aid measures**

## Description of necessary first-aid measures

#### If inhaled

Move the victim into fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration and consult a doctor immediately. Do not use mouth to mouth resuscitation if the victim ingested or inhaled the chemical.

## Following skin contact

Take off contaminated clothing immediately. Wash off with soap and plenty of water. Consult a doctor.

## Following eye contact

Rinse with pure water for at least 15 minutes. Consult a doctor.

#### Following ingestion

Rinse mouth with water. Do not induce vomiting. Never give anything by mouth to an unconscious person. Call a doctor or Poison Control Center immediately.

## Most important symptoms/effects, acute and delayed

no data available

## Indication of immediate medical attention and special treatment needed, if necessary

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR if necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on the left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. Poisons A and B

## **SECTION 5: Firefighting measures**

### Suitable extinguishing media

Advice for firefighters: Wear self-contained breathing apparatus for firefighting if necessary. Tridecafluorohexane-1-sulfonic acid potassium salt

## Specific hazards arising from the chemical

no data available

## Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

## **SECTION 6: Accidental release measures**

## Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

## **Environmental precautions**

Prevent further spillage or leakage if it is safe to do so. Do not let the chemical enter drains. Discharge into the environment must be avoided.

## Methods and materials for containment and cleaning up

Methods and materials for containment and cleaning up: Pick up and arrange disposal without creating dust. Sweep up and shovel. Keep in suitable, closed containers for disposal. Tridecafluorohexane-1-sulfonic acid potassium salt

## **SECTION 7: Handling and storage**

### Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

## Conditions for safe storage, including any incompatibilities

Conditions for safe storage, including any incompatibilities: Keep container tightly closed in a dry and well-ventilated place. Tridecafluorohexane-1-sulfonic acid potassium salt

## SECTION 8: Exposure controls/personal protection

## Control parameters

## Occupational Exposure limit values

no data available

## Biological limit values

no data available

### Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

## Individual protection measures, such as personal protective equipment (PPE)

## Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

## Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

## Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

#### Thermal hazards

no data available

## SECTION 9: Physical and chemical properties and safety characteristics

Physical state: no data available

Colour: no data available

Odour: no data available

Melting no data available

point/freezing

point:

Boiling point or 238-239 deg C

initial boiling point and boiling range:

Flammability: no data available

Lower and upper

no data available

explosion

limit/flammability

limit:

Flash point: no data available

Auto-ignition

no data available

temperature:

Decomposition

no data available

temperature:

pH: no data available

Kinematic no data available

viscosity:

Solubility: In water, 6.2 mg/L at 25 deg C (est)

Partition log Kow = 3.16 (est)

coefficient n-

octanol/water:

Vapour pressure: 0.0046 mm Hg at 25 deg C (est)

Density and/or relative density:

1.841g/cm3

Relative vapour

no data available

density:

Particle

no data available

characteristics:

## **SECTION 10: Stability and reactivity**

## Reactivity

no data available

## Chemical stability

Chemical stability: Stable under recommended storage conditions. Tridecafluorohexane-1-sulfonic acid potassium salt

## Possibility of hazardous reactions

no data available

### Conditions to avoid

no data available

## Incompatible materials

Incompatible materials: Strong oxidizing agents. Tridecafluorohexane-1-sulfonic acid potassium salt

## Hazardous decomposition products

Special hazards arising from the substance or mixture: Carbon oxides, Sulphur oxides, Hydrogen fluoride, Potassium oxides. Tridecafluorohexane-1-sulfonic acid potassium salt

# **SECTION 11: Toxicological information**

## Acute toxicity

Oral: no data available

Inhalation: no data available

Dermal: no data available

## Skin corrosion/irritation

no data available

## Serious eye damage/irritation

no data available

## Respiratory or skin sensitization

no data available

## Germ cell mutagenicity

no data available

## Carcinogenicity

no data available

### Reproductive toxicity

no data available

### STOT-single exposure

no data available

## STOT-repeated exposure

no data available

## Aspiration hazard

no data available

# **SECTION 12: Ecological information**

## **Toxicity**

Toxicity to fish: no data available

Toxicity to daphnia and other aquatic invertebrates: no data available

Toxicity to algae: no data available

Toxicity to microorganisms: no data available

### Persistence and degradability

Biodegradation data specific to perfluorohexanesulfonic acid were not available (SRC, 2015). However, available screening test results indicate that perfluoroalkyl compounds are resistant to biodegradation(1).

## Bioaccumulative potential

A BCF of 10 has been reported for perfluorohexanesulfonic acid(1). Using monitoring data from creek and pond locations in Toronto, Ontario, Canada, a field-based BAF (bioaccumulation factor) of 70 was determined for perfluorohexanesulfonic acid in fish(2). According to a classification scheme(3), these BCF and BAF values suggests the potential for bioconcentration in aquatic organisms is low to moderate(SRC). In living organisms, perfluoroalkyls bind to protein albumin in blood, liver, and eggs and do not accumulate in fat tissue, which may explain why bioconcentration factors (BCFs) are lower than expected in aquatic organisms(4).

## Mobility in soil

Using monitoring data from creek and pond locations in Toronto, Ontario, Canada, a field-based Koc of 9.3 was determined for perfluorohexanesulfonic acid(1). According to a classification scheme(2), this estimated Koc value suggests that perfluorohexanesulfonic acid is expected to have very high mobility in soil. The estimated pKa of perfluorohexanesulfonic acid is 0.14(3), indicating that this compound will exist almost entirely in anion form in the environment and anions generally do not adsorb more strongly to soils containing organic carbon and clay than their neutral counterparts(4). However, it has been suggested that compounds like the analogous PFOS (perfluorooctanesulfonic acid) may form strong bonds via a chemisorption mechanism(5).

#### Other adverse effects

no data available

## **SECTION 13: Disposal considerations**

## Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

## Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

#### **UN Number**

ADR/RID: no data available IMDG: no data available IATA: no data available

## **UN Proper Shipping Name**

ADR/RID: no data available IMDG: no data available

IATA: no data available

## Transport hazard class(es)

ADR/RID: no data available IMDG: no data available IATA: no data available

## Packing group, if applicable

ADR/RID: no data available IMDG: no data available IATA: no data available

#### **Environmental hazards**

ADR/RID: No IMDG: No IATA: No

## Special precautions for user

no data available

## Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

# **EC Inventory**

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Not Listed.

New Zealand Inventory of Chemicals (NZIoC)

Not Listed.

(PICCS)

Not Listed.

Vietnam National Chemical Inventory

Not Listed.

IECSC)

Not Listed.

Korea Existing Chemicals List (KECL)

Not Listed.

## **SECTION 16: Other information**

## Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website:

http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

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