# Chemical Safety Data Sheet MSDS / SDS

## Naphthalene SDS

Revision Date: 2024-04-25 Revision Number: 1

Section 2 Section 3 Section 1 Section 4 Section 5 Section 6 Section 7 Section 8 Section 9 Section 10 Section 11 Section 12 Section 13 Section 14 Section 15 Section 16

## SECTION 1: Identification of the substance/mixture and of the company/undertaking

#### Product identifier

Product name: Naphthalene

CAS: 91-20-3

### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified For R&D use only. Not for medicinal, household or other use.

uses:

Uses advised none

against:

## Company Identification

Company: Chemicalbook.in

Address: 5 vasavi Layout Basaveswara Nilayam Pragathi Nagar Hyderabad, India -500090

Telephone: +91 9550333722

## **SECTION 2: Hazards identification**

### Classification of the substance or mixture

Acute toxicity - Category 4, Oral Carcinogenicity, Category 2

Hazardous to the aquatic environment, short-term (Acute) - Category Acute 1 Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 1

## GHS label elements, including precautionary statements

Pictogram(s)





Signal word Warning

## Hazard statement(s)

H302 Harmful if swallowed H351 Suspected of causing cancer H410 Very toxic to aquatic life with long lasting effects

### Precautionary statement(s)

#### Prevention

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P203 Obtain, read and follow all safety instructions before use.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

P273 Avoid release to the environment.

### Response

P301+P317 IF SWALLOWED: Get medical help.

P330 Rinse mouth.

P318 IF exposed or concerned, get medical advice.

P391 Collect spillage.

## Storage

P405 Store locked up.

## Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

#### Other hazards which do not result in classification

## **SECTION 3: Composition/information on ingredients**

#### Substance

Chemical name: Naphthalene
Common names and Naphthalene

synonyms:

CAS number: 91-20-3
EC number: 202-049-5
Concentration: 100%

#### **SECTION 4: First aid measures**

### Description of necessary first-aid measures

#### If inhaled

Fresh air. Refer for medical attention.

### Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap. Seek medical attention if you feel unwell.

### Following eye contact

Rinse with plenty of water (remove contact lenses if easily possible).

## Following ingestion

Rinse mouth. Do NOT induce vomiting. Refer immediately for medical attention.

## Most important symptoms/effects, acute and delayed

Excerpt from ERG Guide 133 [Flammable Solids]: Fire may produce irritating and/or toxic gases. Contact may cause burns to skin and eyes. Contact with molten substance may cause severe burns to skin and eyes. Runoff from fire control may cause pollution. (ERG, 2016)

### Indication of immediate medical attention and special treatment needed, if necessary

Emergency and supportive measures: Maintain an open airway and assist ventilation if necessary. Treat coma and seizures if they occur. Treat hemolysis and resulting hemoglobinuria if they occur, by intravenous hydration and urinary alkalinization. Naphthalene and paradichlorobenzene

## **SECTION 5: Firefighting measures**

### Suitable extinguishing media

If material on fire or involved in fire: Do not extinguish fire unless flow can be stopped. Use water in flooding quantities as fog. Cool all affected containers with flooding quantities of water. Apply water from as far a distance as possible. Solid streams of water may be ineffective. Use "alcohol" foam, dry chemical or carbon dioxide. Naphthalene, crude or refined

### Specific hazards arising from the chemical

Excerpt from ERG Guide 133 [Flammable Solids]: Flammable/combustible material. May be ignited by friction, heat, sparks or flames. Some may burn rapidly with flare-burning effect. Powders, dusts, shavings, borings, turnings or cuttings may explode or burn with explosive violence. Substance may be transported in a molten form at a temperature that may be above its flash point. May re-ignite after fire is extinguished. (ERG, 2016)

### Special protective actions for fire-fighters

Use water spray, powder, foam, carbon dioxide.

### **SECTION 6: Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Do NOT wash away into sewer. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

## **Environmental precautions**

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Do NOT let this chemical enter the environment. Do NOT wash away into sewer. Sweep spilled substance into covered containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

### Methods and materials for containment and cleaning up

Environmental considerations: Water spill: Use natural barriers or oil spill booms to limit spill travel. Use natural deep water pockets, excavated lagoons, or sand bag barriers to trap material at bottom. Remove trapped material with suction hoses. Use mechanical dredges or lifts to remove immobilized masses of pollutants and precipitates. Naphthalene, molten

## **SECTION 7: Handling and storage**

## Precautions for safe handling

NO open flames. Closed system, dust explosion-proof electrical equipment and lighting. Prevent deposition of dust. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### Conditions for safe storage, including any incompatibilities

Separated from strong oxidants and food and feedstuffs. Store in an area without drain or sewer access. Provision to contain effluent from fire extinguishing. Without inert-gas blanketing and at the temperature normally used for the storage of molten naphthalene, i.e., 90 deg C, the vapors above the liquid are within the flammability limits. Thus, storage tanks containing molten naphthalene have a combustible mixture in the vapor space and care must be taken to eliminate all sources of ignition around such systems. Naphthalene dust can form explosive mixtures with air which necessitates the design and operation of solid handling systems.

## SECTION 8: Exposure controls/personal protection

## Control parameters

## Occupational Exposure limit values

TLV: 10 ppm as TWA; (skin); A3 (confirmed animal carcinogen with unknown relevance to humans). EU-OEL: 50 mg/m3, 10 ppm as TWA.MAK: skin absorption (H); carcinogen category: 2; germ cell mutagen group: 3B

## Biological limit values

no data available

## Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

## Individual protection measures, such as personal protective equipment (PPE)

## Eye/face protection

Wear safety spectacles.

### Skin protection

Protective gloves. Protective clothing.

## Respiratory protection

Use ventilation (not if powder), local exhaust or breathing protection.

#### Thermal hazards

no data available

## SECTION 9: Physical and chemical properties and safety characteristics

Physical state: Solid. Flakes, granules.

kPa.

Colour: White as solid.

Odour: Aromatic odor

Melting 80.3 °C. Atm. press.:Ca. 101 kPa. Remarks:Pure naphthalene.;78.9 °C. Atm. press.:Ca. 101

point/freezing

•

point:

Boiling point or 218.1 °C. Atm. press.:101.3 kPa. Remarks:Naphthalene (pure).;218.2 °C. Atm. press.:101.3

initial boiling point and boiling range:

kPa. Remarks: Naphthalene (techn.).

Flammability: Combustible Solid, but will take some effort to ignite.

Lower and upper

Lower flammable limit: 0.9% by volume; Upper flammable limit: 5.9% by volume

explosion

limit/flammability

limit:

Flash point: 78.5 °C. Atm. press.:99 kPa.

Auto-ignition >= 526 - <= 587 °C. Atm. press.:1 013 hPa. Remarks:Range of literature data;540 °C. Atm.

temperature: press.:1 013 hPa. Remarks: Nabert and Schon 1963: DIN 51794.

Decomposition

no data available

temperature:

pH: no data available

Kinematic viscosity:

kinematic viscosity (in mm2/s) = 1.05. Temperature:81.5°C. Remarks:Naphthalene, pure.;kinematic viscosity (in mm2/s) = 1.03. Temperature:85.0°C. Remarks:Naphthalene,

pure.;kinematic viscosity (in mm2/s) = 1. Temperature:80.0°C. Remarks:Naphthalene,

technical quality.

Solubility: Insoluble in water

Partition coefficient n-

log Pow = 3.4. Temperature:25 °C.

coefficient noctanol/water:

Vapour pressure: 7.2 Pa. Temperature: 20 °C.; 10.5 Pa. Temperature: 25 °C.

Density and/or 1.085. Temperature:24.7 °C.;1.085. Temperature:20 °C.;1.069 g/cm3. Temperature:24.7

relative density: °C.

Relative vapour

density:

4.4 (vs air)

Particle no data available

characteristics:

## **SECTION 10: Stability and reactivity**

### Reactivity

On combustion, forms irritating and toxic gases. Reacts with strong oxidants. This generates fire and explosion hazard.

## Chemical stability

Stable under recommended storage conditions.

### Possibility of hazardous reactions

Flammable in the presence of a source of ignition, through friction or retained heat. Dust explosion possible if in powder or granular form, mixed with air. Vigorous reactions, sometimes amounting to explosions, can result from the contact between aromatic hydrocarbons, such as NAPHTHALENE, and strong oxidizing agents. They can react exothermically with bases and with diazo compounds. Substitution at the benzene nucleus occurs by halogenation (acid catalyst), nitration, sulfonation, and the Friedel-Crafts reaction. Naphthalene, camphor, glycerol, or turpentine will react violently with chromic anhydride [Haz. Chem.

Data 1967. p 68]. Friedel-Crafts acylation of naphthalene using benzoyl chloride, catalyzed by AlCl3, must be conducted above the melting point of the mixture, or the reaction may be violent [Clar, E. et al., Tetrahedron, 1974, 30, 3296].

#### Conditions to avoid

no data available

## Incompatible materials

Naphthalene ...will react violently with chromic anhydride.

## Hazardous decomposition products

When heated to decomposition it emits acrid smoke and irritating fumes.

## **SECTION 11: Toxicological information**

### Acute toxicity

Oral: LD50 - mouse (female) - 710 mg/kg bw.

Inhalation: LC50 - rat (male/female) - > 77.7 ppm.

Dermal: LD50 - rat (male/female) - > 16 000 mg/kg bw.

#### Skin corrosion/irritation

no data available

## Serious eye damage/irritation

no data available

## Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

## Carcinogenicity

WEIGHT-OF-EVIDENCE CHARACTERIZATION: Using criteria of the 1986 Guidelines for Carcinogen Risk Assessment, naphthalene is classified in group C, a possible human carcinogen. This is based on the inadequate data of carcinogenicity in humans exposed to naphthalene via the oral and inhalation routes, and the limited evidence of carcinogenicity in animals via the inhalation route. Using the 1996 Proposed Guidelines for Carcinogen Risk Assessment, the human carcinogenic potential of naphthalene via the oral or inhalation routes "cannot be determined" at this time based on human and animal data; however, there is suggestive evidence (observations of benign respiratory tumors and one carcinoma in female mice only exposed to naphthalene by inhalation). Additional support includes increase in respiratory tumors associated with exposure to 1-methylnaphthalene. At the present time the mechanism whereby naphthalene produces benign respiratory tract tumors are not fully understood, but are hypothesized to involve oxygenated reactive metabolites produced via the cytochrome P-450 monooxygenase system. However, based on the many negative results obtained in genotoxicity tests, a genotoxic mechanism appears unlikely. HUMAN CARCINOGENICITY DATA: Available data are inadequate to establish a causal association between exposure to naphthalene and cancer in humans. Adequately scaled epidemiological studies designed to examine a possible association between naphthalene exposure and cancer were not located. Overall, no data are available to evaluate the carcinogenic potential in exposed human populations.

### Reproductive toxicity

Hemolytic anemia has been reported in infants born to mothers who "sniffed" and ingested naphthalene (as mothballs) during pregnancy. The mothers themselves were anemic, but to a lesser extent than the infants. Signs of maternal toxicity (e.g., decreased body weight and lethargy) but no fetal effects were reported in rats and rabbits exposed to naphthalene via gavage. Maternal toxicity (increased mortality and reduced weight gain) and fetotoxicity (reduced number of live pups per litter) were observed in mice exposed via gavage.

## STOT-single exposure

The substance may cause effects on the blood. This may result in lesions of blood cells (haemolysis). See Notes. The effects may be delayed. Ingestion could cause death. Medical observation is indicated.

## STOT-repeated exposure

The substance may have effects on the blood. This may result in chronic haemolytic anaemia. The substance may have effects on the eyes. This may result in development of cataract. This substance is possibly carcinogenic to humans.

## Aspiration hazard

A harmful contamination of the air will be reached rather slowly on evaporation of this substance at 20°C.

#### **Toxicity**

Toxicity to fish: LC50 - Oncorhynchus mykiss (previous name: Salmo gairdneri) - 1.6 mg/L - 96 h. Remarks: Naphthalene.

Toxicity to daphnia and other aquatic invertebrates: EC50 - Daphnia magna - 2.16 mg/L - 48 h.

Toxicity to algae: EC50 - Pseudokirchneriella subcapitata (previous names: Raphidocelis subcapitata, Selenastrum capricornutum) - 2.96 mg/L - 4 h.

Toxicity to microorganisms: IC50 for Nitrosomonas - Nitrosomonas and aerobic heterotrophs - 29 mg/L - 24 h.

### Persistence and degradability

AEROBIC: Although there are some conflicting data, data suggest that naphthalene degrades after a relatively short period of acclimation and that degradation can be rapid in oil polluted water, slow in unpolluted water and that the rate of degradation increases with the concentration of naphthalene(1). In laboratory tests with sewage or sludge inoculums, 100% degradation was obtained in 7 days(2-3) while others got 0% BOD in 5 days(4-5). The lag period for naphthalene degradation decreased as groundwater was more polluted with fuel oil; the lag period was 1.2 and 1.9 days in heavily polluted and slightly polluted water, respectively versus 12 days for unpolluted water(6). Approximately 70% of naphthalene was lost in a pilot-scale municipal wastewater treatment plant due to biodegradation(7). In water, bacteria can utilize naphthalene only when it is in the dissolved state(8). Naphthalene, present at 100 mg/L, reached 2% of its theoretical BOD in 4 weeks using an activated sludge inoculum at 30 mg/L and the Japanese MITI test(9).

## Bioaccumulative potential

The BCF of naphthalene at a water concentration of 0.15 mg/L ranged from 36.5 to 168 in carp, over an 8 week exposure period(1). BCF values in sheepshead minnows (Cyprinodon variegatus) exposed for 36 days to 1.34 and 12.5 ug/L of naphthalene were 692 and 714, respectively(2). According to a classification scheme(3), these BCF values suggest for bioconcentration in aquatic organisms is low to high(SRC). BCF values in amphipods (Diporeia spp) exposed for 28 days to 453.2 to 2201.1 ug/L of naphthalene were 490 to 736(4).

## Mobility in soil

The mean Koc value of naphthalene in 17 soils throughout the US was 871(1). The Koc of naphthalene in 5 different soils from Japan ranged from 440-830(2). In a long term field experiment in Uppsala, Sweden, soils were treated with 7 types of amendments and 1 untreated plot, Koc values were 112-680 for naphthalene(3). In a contaminated soil collected from a railway station in Spain, naphthalene had a log Koc of 3.97(4). The mean Koc value of naphthalene in sediment samples from 2 ponds and 1 river from north Georgia, US was 1,300(5). The log Koc value in sediment was 2.84(6). In 76 sediment samples from 7 sites in New York and 1 in North Carolina, naphthalene had log Koc values of 2.45-5.59(7). According to a classification scheme(8), these Koc values suggest that naphthalene is expected to have high to no mobility in soil(SRC).

#### Other adverse effects

no data available

## **SECTION 13: Disposal considerations**

## Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

## Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

## **SECTION 14: Transport information**

#### **UN Number**

ADR/RID: UN1334 (For reference only, please check.) IMDG: UN1334 (For reference only, please check.) IATA: UN1334 (For reference only, please check.)

## **UN Proper Shipping Name**

ADR/RID: NAPHTHALENE, CRUDE or NAPHTHALENE, REFINED (For reference only, please check.)

IMDG: NAPHTHALENE, CRUDE or NAPHTHALENE, REFINED (For reference only, please check.)

IATA: NAPHTHALENE, CRUDE or NAPHTHALENE, REFINED (For reference only, please check.)

## Transport hazard class(es)

ADR/RID: 4.1 (For reference only, please check.) IMDG: 4.1 (For reference only, please check.) IATA: 4.1 (For reference only, please check.)

## Packing group, if applicable

ADR/RID: III (For reference only, please check.)
IMDG: III (For reference only, please check.)
IATA: III (For reference only, please check.)

### **Environmental hazards**

ADR/RID: Yes IMDG: Yes IATA: Yes

## Special precautions for user

no data available

## Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

**EC Inventory** 

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Listed.

New Zealand Inventory of Chemicals (NZIoC)

Listed.

(PICCS)

Listed.

## Vietnam National Chemical Inventory

Listed.

IECSC)

Listed.

Korea Existing Chemicals List (KECL)

Listed.

#### **SECTION 16: Other information**

#### Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website:

http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

Disclaimer: The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. We as supplier shall not be held liable for any