# Chemical Safety Data Sheet MSDS / SDS

## Mercury(II) sulfide SDS

Revision Date: 2024-04-25 Revision Number: 1

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## SECTION 1: Identification of the substance/mixture and of the company/undertaking

#### Product identifier

Product name: Mercury(II) sulfide

CAS: 1344-48-5

### Relevant identified uses of the substance or mixture and uses advised against

Relevant identified For R&D use only. Not for medicinal, household or other use.

uses:

Uses advised

against:

## Company Identification

Company: Chemicalbook.in

none

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### **SECTION 2: Hazards identification**

### Classification of the substance or mixture

Skin sensitization, Category 1

### GHS label elements, including precautionary statements

Pictogram(s)

**(** 

Signal word

Warning

## Hazard statement(s)

H317 May cause an allergic skin reaction

### Precautionary statement(s)

#### Prevention

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P272 Contaminated work clothing should not be allowed out of the workplace.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

### Response

P302+P352 IF ON SKIN: Wash with plenty of water/...

P333+P317 If skin irritation or rash occurs: Get medical help.

P321 Specific treatment (see ... on this label).

P362+P364 Take off contaminated clothing and wash it before reuse.

### Storage

none

### Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

### Other hazards which do not result in classification

no data available

## **SECTION 3: Composition/information on ingredients**

#### Substance

Chemical name: Mercury(II) sulfide

Common names and

Mercury(II) sulfide

synonyms:

CAS number: 1344-48-5 EC number: 215-696-3

Concentration: 100%

#### **SECTION 4: First aid measures**

### Description of necessary first-aid measures

#### If inhaled

Move the victim into fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration and consult a doctor immediately. Do not use mouth to mouth resuscitation if the victim ingested or inhaled the chemical.

### Following skin contact

Take off contaminated clothing immediately. Wash off with soap and plenty of water. Consult a doctor.

### Following eye contact

Rinse with pure water for at least 15 minutes. Consult a doctor.

### Following ingestion

Rinse mouth with water. Do not induce vomiting. Never give anything by mouth to an unconscious person. Call a doctor or Poison Control Center immediately.

### Most important symptoms/effects, acute and delayed

Acute poisoning can result from inhaling dust concentrations of 1.2-8.5 mg/m 3 in air; symptoms include pain and tightness in chest, coughing, and difficulty in breathing. If ingested, toxicity depends on release of the Hg ++ ion; chronic mercury poisoning can cause kidney, mental, and nervous disturbances. Dust irritates eyes and frequently causes allergic dermatitis; absorption through skin can cause systemic poisoning. (USCG, 1999)

### Indication of immediate medical attention and special treatment needed, if necessary

no data available

## **SECTION 5: Firefighting measures**

### Suitable extinguishing media

Fire Extinguishing Agents Not to Be Used: Other agents may be ineffective. Fire Extinguishing Agents: Water, foam, sand (USCG, 1999)

### Specific hazards arising from the chemical

Special Hazards of Combustion Products: Smoke from fire contains poisonous mercury vapor and irritating sulfur dioxide gas. Behavior in Fire: Changes color when hot. Decomposes at burning temperature. The black form may soften, and molten sulfur may flow out and burn. (USCG, 1999)

### Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

#### **SECTION 6: Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

## Environmental precautions

Prevent further spillage or leakage if it is safe to do so. Do not let the chemical enter drains. Discharge into the environment must be avoided.

### Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

### **SECTION 7: Handling and storage**

### Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

## Conditions for safe storage, including any incompatibilities

Store the container tightly closed in a dry, cool and well-ventilated place. Store apart from foodstuff containers or incompatible materials.

## SECTION 8: Exposure controls/personal protection

## Control parameters

### Occupational Exposure limit values

| Component | Mercury(II) sulfide       |                   |                          |                   |  |
|-----------|---------------------------|-------------------|--------------------------|-------------------|--|
| CAS No.   | 1344-48-5                 |                   |                          |                   |  |
|           | Limit value - Eight hours |                   | Limit value - Short term |                   |  |
|           | ppm                       | <sub>mg/m</sub> 3 | ppm                      | <sub>mg/m</sub> 3 |  |
| Finland   | ?                         | 0,02 (1)          | ?                        | ?                 |  |
|           | Remarks                   |                   |                          |                   |  |
| Finland   | (1) calculated as Hg      |                   |                          |                   |  |

## Biological limit values

no data available

### Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

### Individual protection measures, such as personal protective equipment (PPE)

## Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

### Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands.

The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

## Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

#### Thermal hazards

no data available

## SECTION 9: Physical and chemical properties and safety characteristics

Physical state: Mercuric sulfide is an odorless red or black solid. Sinks in water. Insoluble in water.

Colour: no data available

Odour: no data available

Melting 584°C

point/freezing

point:

Boiling point or 584°C

initial boiling point and boiling range:

Flammability: no data available

Lower and upper no data available

explosion

limit/flammability

limit:

Flash point: no data available

Auto-ignition

no data available

temperature:

**Decomposition** no data available

temperature:

pH: no data available

Kinematic no dat

viscosity:

no data available

Solubility: no data available

Partition no data available

coefficient noctanol/water:

Vapour pressure: no data available

Density and/or 8.1 g/mL at 25°C

relative density:

Relative vapour no data available

density:

Particle no data available

characteristics:

## **SECTION 10: Stability and reactivity**

### Reactivity

Readily oxidized as to be pyrophoric in air [Bretherick 1979. p. 120].

## Chemical stability

no data available

## Possibility of hazardous reactions

Inorganic reducing agents, such as MERCURIC SULFIDE, react with oxidizing agents to generate heat and products that may be flammable, combustible, or otherwise reactive.

## Conditions to avoid

no data available

## Incompatible materials

no data available

## Hazardous decomposition products

# **SECTION 11: Toxicological information**

## Acute toxicity

Oral: no data available

Inhalation: no data available

Dermal: no data available

#### Skin corrosion/irritation

no data available

## Serious eye damage/irritation

no data available

## Respiratory or skin sensitization

no data available

## Germ cell mutagenicity

no data available

## Carcinogenicity

no data available

# Reproductive toxicity

no data available

## STOT-single exposure

no data available

## STOT-repeated exposure

no data available

## Aspiration hazard

no data available

# **SECTION 12: Ecological information**

## **Toxicity**

Toxicity to fish: no data available

Toxicity to daphnia and other aquatic invertebrates: no data available

Toxicity to algae: no data available

Toxicity to microorganisms: no data available

## Persistence and degradability

no data available

## Bioaccumulative potential

no data available

## Mobility in soil

no data available

#### Other adverse effects

no data available

# **SECTION 13: Disposal considerations**

## Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

## Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

## **SECTION 14: Transport information**

#### **UN Number**

ADR/RID: UN2025 (For reference only, please check.) IMDG: UN2025 (For reference only, please check.) IATA: UN2025 (For reference only, please check.)

### **UN Proper Shipping Name**

ADR/RID: MERCURY COMPOUND, SOLID, N.O.S. (For reference only, please check.) IMDG: MERCURY COMPOUND, SOLID, N.O.S. (For reference only, please check.) IATA: MERCURY COMPOUND, SOLID, N.O.S. (For reference only, please check.)

## Transport hazard class(es)

ADR/RID: 6.1 (For reference only, please check.)
IMDG: 6.1 (For reference only, please check.)
IATA: 6.1 (For reference only, please check.)

## Packing group, if applicable

ADR/RID: I (For reference only, please check.)
IMDG: I (For reference only, please check.)
IATA: I (For reference only, please check.)

### Environmental hazards

ADR/RID: No IMDG: No IATA: No

## Special precautions for user

no data available

## Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

**EC Inventory** 

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Listed.

New Zealand Inventory of Chemicals (NZIoC)

Listed.

(PICCS)

Listed.

Vietnam National Chemical Inventory

Listed.

IECSC)

Listed.

#### Korea Existing Chemicals List (KECL)

Listed.

#### **SECTION 16: Other information**

### Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website:

http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

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