# Chemical Book India

MC		Chem	ical Safety	Data Shee	t MSDS / S	DS	The second s
			Dibro Revision Date:20	momethane SI 24-04-25 Revision	<b>DS</b> n Number:1		
Section 1 Section 9	Section 2 Section 10	Section 3 Section 11	Section 4 Section 12	Section 5 Section 13	Section 6 Section 14	Section 7 Section 15	Section 8 Section 16
SECTION 1: Identification of the substance/mixture and of the company/undertaking Product identifier							
Product name:		Dibromomethane					
CAS:		74-95-3					
Relevant ide	entified uses o	of the substance	or mixture and	l uses advised a	gainst		
Relevant ide uses:	entified	For R&D use only.	Not for medici	nal, household o	or other use.		
Uses advised against:	b	none					
Company Id	lentification						
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# **SECTION 2: Hazards identification**

## Classification of the substance or mixture

Acute toxicity - Category 4, Inhalation Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 3

#### GHS label elements, including precautionary statements

Pictogram(s)

Signal word

Warning

#### Hazard statement(s)

H332 Harmful if inhaled H412 Harmful to aquatic life with long lasting effects

#### Precautionary statement(s)

#### Prevention

P261 Avoid breathing dust/fume/gas/mist/vapours/spray. P271 Use only outdoors or in a well-ventilated area. P273 Avoid release to the environment.

#### Response

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing. P317 Get medical help.

#### Storage

none

#### Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

#### Other hazards which do not result in classification

no data available

# SECTION 3: Composition/information on ingredients Substance

Chemical name:	Dibromomethane
Common names and synonyms:	Dibromomethane
CAS number:	74-95-3
EC number:	200-824-2
Concentration:	100%

# **SECTION 4: First aid measures**

#### Description of necessary first-aid measures

#### If inhaled

Fresh air, rest. Administration of oxygen may be needed. Artificial respiration may be needed. Refer for medical attention.

#### Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap.

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

#### Following ingestion

Rinse mouth. Do NOT induce vomiting. Refer for medical attention .

#### Most important symptoms/effects, acute and delayed

INHALATION: Anesthetic effects, nausea and drunkenness. CONTACT WITH SKIN AND EYES: Skin irritation of eyes and nose. (USCG, 1999)

#### Indication of immediate medical attention and special treatment needed, if necessary

Basic treatment: Establish a patent airway. Suction if necessary. Watch for signs of respiratory insufficiency and assist ventilations as needed. Administer oxygen by nonrebreather mask at 10 to 15 L/min. Minimize physical activity and provide a quiet atmosphere. Monitor for pulmonary edema and treat if necessary. Anticipate seizures and treat if necessary. For eye contamination, flush eyes immediately with water. Irrigate each eye continuously with normal saline during transport. Do not use emetics. Rinse mouth and administer 5 ml/kg up to 200 ml of water for dilution if the patient can swallow, has a strong gag reflex, and does not drool. Administer activated charcoal. Treat frostbite with rapid rewarming techniques. Chlorinated fluorocarbons

# **SECTION 5: Firefighting measures**

## Suitable extinguishing media

If material involved in fire: Extinguish fire using agent suitable for type of surrounding fire. (Material itself does not burn or burns with difficulty.) Use foam, dry chemical, or carbon dioxide.

## Specific hazards arising from the chemical

Special Hazards of Combustion Products: Dissociation products generated in a fire may be irritating or toxic. (USCG, 1999)

## Special protective actions for fire-fighters

In case of fire in the surroundings, use appropriate extinguishing media. In case of fire: keep drums, etc., cool by spraying with water.

# SECTION 6: Accidental release measures

#### Personal precautions, protective equipment and emergency procedures

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Ventilation. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

#### Environmental precautions

Personal protection: filter respirator for organic gases and vapours adapted to the airborne concentration of the substance. Ventilation. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

## Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

# SECTION 7: Handling and storage

#### Precautions for safe handling

See Chemical Dangers Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

#### Conditions for safe storage, including any incompatibilities

Separated from food and feedstuffs, strong oxidants, strong bases and metals. Do NOT store or transport in containers made from aluminium or plastic. Ventilation along the floor.... MATERIALS WHICH ARE TOXIC AS STORED OR WHICH CAN DECOMP INTO TOXIC COMPONENTS DUE TO CONTACT WITH HEAT, MOISTURE, ACID, OR ACID FUMES, SHOULD BE STORED IN COOL, WELL VENTILATED PLACE, OUT OF DIRECT RAYS OF SUN, AWAY FROM AREAS OF HIGH FIRE HAZARD & SHOULD BE PERIODICALLY INSPECTED & MONITORED.

# SECTION 8: Exposure controls/personal protection

#### **Control parameters**

#### Occupational Exposure limit values

Component	Dibromometha	Dibromomethane				
CAS No.	74-95-3	74-95-3				
	Limit value - E	Eight hours	Limit value -	Limit value - Short term		
	ppm	<sub>mg/m</sub> 3	ppm	<sub>mg/m</sub> 3		
Latvia	?	10	?	?		
	Remarks					

#### **Biological limit values**

no data available

#### Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

#### Individual protection measures, such as personal protective equipment (PPE)

## Eye/face protection

Wear safety spectacles or eye protection in combination with breathing protection.

# Skin protection

Protective gloves. Protective clothing.

# Respiratory protection

Use ventilation, local exhaust or breathing protection.

# Thermal hazards

no data available

# SECTION 9: Physical and chemical properties and safety characteristics

Physical state:	Dibromomethane is a colorless liquid with a pleasant odor. Insoluble in water and denser than water. May be toxic by ingestion. Used as a solvent and as a motor fuel.	
Colour:	Clear, colorless liquid	
Odour:	no data available	
Melting point/freezing point:	120°C(dec.)(lit.)	
Boiling point or initial boiling point and boiling range:	97°C	
Flammability:	Not combustible. Gives off irritating or toxic fumes (or gases) in a fire.	
Lower and upper explosion limit/flammability limit:	no data available	
Flash point:	-10°C(lit.)	
Auto-ignition temperature:	515°C	
Decomposition temperature:	no data available	
pH:	no data available	

Kinematic viscosity:	1.320 mPa @ 0 deg C; 0.980 mPa @ 25 deg C
Solubility:	11.70 g/1000 g water @ 15 deg C; 11.93 g/1000 g water @ 30 deg C
Partition coefficient n- octanol/water:	log Kow = 1.70
Vapour pressure:	34.9 mm Hg ( 20 °C)
Density and/or relative density:	2.477
Relative vapour density:	6.05 (vs air)
Particle characteristics:	no data available

# SECTION 10: Stability and reactivity

#### Reactivity

Decomposes on heating. This produces toxic and corrosive fumes including hydrogen bromide and carbon monoxide (see ICSC 0023). Reacts violently with strong oxidants, strong bases and some metals such as aluminium, magnesium and potassium. This generates fire and explosion hazard. Attacks some forms of plastic, rubber and coatings.

#### Chemical stability

no data available

#### Possibility of hazardous reactions

NOT FLAWWABLE BY STANDARD TEST IN AIRThe vapour is heavier than air. Halogenated aliphatic compounds, such as DIBROMOMETHANE, are moderately or very reactive. Halogenated organics generally become less reactive as more of their hydrogen atoms are replaced with halogen atoms. Low molecular weight haloalkanes are highly flammable and can react with some metals to form dangerous products. Materials in this group are incompatible with strong oxidizing and reducing agents. Also, they are incompatible with many amines, nitrides, azo/diazo compounds, alkali metals (potassium), and epoxides.

## Conditions to avoid

no data available

#### Incompatible materials

Although apparently stable on contact, mixtures of potassium (or its alloys) with a wide range of halocarbons are shock-sensitive & may explode with great violence on light impact. Chloroethane, dichloroethane ... dibromomethane & diiodomethane are among those investigated. Sensitivity increases generally with the degree of substitution ...

#### Hazardous decomposition products

When heated to decomp it emits toxic fumes of /hydrogen bromide/.

# SECTION 11: Toxicological information

Acute toxicity Oral: no data available

Inhalation: no data available

Dermal: no data available

#### Skin corrosion/irritation

no data available

#### Serious eye damage/irritation

no data available

#### Respiratory or skin sensitization

no data available

# Germ cell mutagenicity

no data available

# Carcinogenicity

#### Reproductive toxicity

no data available

#### STOT-single exposure

The substance is mildly irritating to the eyes, skin and respiratory tract. If swallowed the substance may cause vomiting and could result in aspiration pneumonitis. The substance may cause effects on the central nervous system, blood and heart. Exposure could cause carbon monoxide poisoning. This may result in impaired functions. The effects may be delayed. Exposure at high concentrations could cause unconsciousness or death.

#### STOT-repeated exposure

Repeated or prolonged contact with skin may cause dermatitis. The substance may have effects on the central nervous system, liver, kidneys and lungs.

#### Aspiration hazard

A harmful contamination of the air can be reached very quickly on evaporation of this substance at 20°C.

# SECTION 12: Ecological information

Toxicity

Toxicity to fish: no data available Toxicity to daphnia and other aquatic invertebrates: no data available Toxicity to algae: no data available Toxicity to microorganisms: no data available

#### Persistence and degradability

no data available

#### Bioaccumulative potential

An estimated BCF of 4.06 was calculated for dibromomethane(SRC), using a log Kow of 1.70(1) and a regression-derived equation(2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organisms is low.

#### Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc for dibromomethane can be estimated to be 24(SRC). According to a classification scheme(2), this estimated Koc value suggests that dibromomethane is expected to have very high mobility in soil.

#### Other adverse effects

no data available

# SECTION 13: Disposal considerations

#### Disposal methods

## Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

## Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

## **UN Number**

ADR/RID: UN2664 (For reference only, please check.) IMDG: UN2664 (For reference only, please check.) IATA: UN2664 (For reference only, please check.)

# **UN Proper Shipping Name**

ADR/RID: DIBROMOMETHANE (For reference only, please check.) IMDG: DIBROMOMETHANE (For reference only, please check.) IATA: DIBROMOMETHANE (For reference only, please check.)

## Transport hazard class(es)

ADR/RID: 6.1 (For reference only, please check.) IMDG: 6.1 (For reference only, please check.) IATA: 6.1 (For reference only, please check.)

## Packing group, if applicable

ADR/RID: III (For reference only, please check.) IMDG: III (For reference only, please check.) IATA: III (For reference only, please check.)

#### Environmental hazards

ADR/RID: No IMDG: No IATA: No

#### Special precautions for user

no data available

#### Transport in bulk according to IMO instruments

no data available

## **SECTION 15: Regulatory information**

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

## EC Inventory

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015 Listed. New Zealand Inventory of Chemicals (NZIoC) Listed. (PICCS) Listed. Vietnam National Chemical Inventory Listed. IECSC) Listed. Korea Existing Chemicals List (KECL) Listed.

# **SECTION 16: Other information**

Abbreviations and acronyms CAS: Chemical Abstracts Service ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road RID: Regulation concerning the International Carriage of Dangerous Goods by Rail IMDG: International Maritime Dangerous Goods IATA: International Air Transportation Association TWA: Time Weighted Average STEL: Short term exposure limit LC50: Lethal Concentration 50% EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=O&request\_locale=en

CAWEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

## Other Information

Do NOT use in the vicinity of a fire or a hot surface, or during welding.

Disclaimer: The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. We as supplier shall not be held liable for any