Chemical Safety Data Sheet MSDS / SDS

Butane-1-thiol SDS

Revision Date: 2024-04-25 Revision Number: 1

Section 2 Section 3 Section 5 Section 6 Section 8 Section 1 Section 4 Section 7 Section 9 Section 10 Section 11 Section 12 Section 13 Section 14 Section 15 Section 16

SECTION 1: Identification of the substance/mixture and of the company/undertaking

Product identifier

Product name: Butane-1-thiol

CAS: 109-79-5

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified For R&D use only. Not for medicinal, household or other use.

uses:

Uses advised none

against:

Company Identification

Company: Chemicalbook.in

Address: 5 vasavi Layout Basaveswara Nilayam Pragathi Nagar Hyderabad, India -500090

Telephone: +91 9550333722

SECTION 2: Hazards identification

Classification of the substance or mixture

Not classified.

GHS label elements, including precautionary statements

Pictogram(s)







Signal word

Danger

Hazard statement(s)

H225 Highly flammable liquid and vapour H302 Harmful if swallowed H317 May cause an allergic skin reaction H410 Very toxic to aquatic life with long lasting effects

Precautionary statement(s)

Prevention

none

Response

none

Storage

none

Disposal

none

Other hazards which do not result in classification

no data available

SECTION 3: Composition/information on ingredients

Substance

Chemical name: Butane-1-thiol
Common names and Butane-1-thiol

synonyms:

CAS number: 109-79-5 EC number: 203-705-3

Concentration: 100%

SECTION 4: First aid measures

Description of necessary first-aid measures

If inhaled

Fresh air, rest. Artificial respiration may be needed. Refer for medical attention.

Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower. Refer for medical attention.

Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

Following ingestion

Rinse mouth. Refer for medical attention.

Most important symptoms/effects, acute and delayed

Inhalation causes loss of sense of smell; muscular weakness. convulsions, and respiratory paralysis may follow prolonged exposure. Contact of liquid with eyes or skin causes slight irritation. Ingestion causes nausea. (USCG, 1999)

Indication of immediate medical attention and special treatment needed, if necessary

Basic treatment: Establish a patent airway. Suction if necessary. Watch for signs of respiratory insufficiency and assist ventilations if necessary. Administer oxygen by nonrebreather mask at 10 to 15 L/min. Monitor for pulmonary edema and treat if necessary . Anticipate seizures and treat if necessary . For eye contamination, flush eyes immediately with water. Irrigate each eye continuously with normal saline during transport . Do not use emetics. For ingestion, rinse mouth and administer 5 m1/kg up to 200 ml of water for dilution if the patient can swallow, has a strong gag reflex, and does not drool . Administer activated charcoal . Cover skin burns with dry sterile dressings after decontamination . Sulfur and related compounds

SECTION 5: Firefighting measures

Suitable extinguishing media

Alcohol foam. Water may be ineffective.

Specific hazards arising from the chemical

Special Hazards of Combustion Products: Irritating sulfur dioxide may form. Behavior in Fire: Vapors are heavier than air and may travel long distance to a source of ignition and flash back. (USCG, 1999)

Special protective actions for fire-fighters

Use alcohol-resistant foam, powder, carbon dioxide. In case of fire: keep drums, etc., cool by spraying with water.

SECTION 6: Accidental release measures

Personal precautions, protective equipment and emergency procedures

Evacuate danger area! Personal protection: chemical protection suit including self-contained breathing apparatus. Remove all ignition sources. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations. Do NOT wash away into sewer.

Environmental precautions

Evacuate danger area! Personal protection: chemical protection suit including self-contained breathing apparatus. Remove all ignition sources. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations. Do NOT wash away into sewer.

Methods and materials for containment and cleaning up

If n-butyl mercaptan is spilled or leaked, the following steps should be taken: 1. Remove all ignition sources. 2. Ventilate area of spill or leak. 3. For small quantities of liquids containing n-butyl mercaptan, absorb on paper towels and place in an appropriate container. Place towels in a safe place such as a fume hood for evaporation. Allow sufficient time for evaporation of the vapors so that the hood ductwork is free from n-butyl mercaptan vapors. Burn the paper in a suitable location away from combustible materials. 4. Large quantities of liquids containing n-butyl mercaptan may be absorbed in vermiculite, dry sand, earth, or a similar material and placed in an appropriate container. n-Butyl mercaptan should not be allowed to enter a confined space such as a sewer because of the possibility of an explosion. 5. Liquids containing n-butyl mercaptan may be collected by vacuuming with an appropriate system. If a vacuum system is used, there should be no sources of ignition in the vicinity of the spill, and flashback prevention devices should be provided.

SECTION 7: Handling and storage

Precautions for safe handling

NO open flames, NO sparks and NO smoking. Closed system, ventilation, explosion-proof electrical equipment and lighting. Do NOT use compressed air for filling, discharging, or handling. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

Conditions for safe storage, including any incompatibilities

Fireproof. Separated from strong oxidants and acids. Fireproof. Separated from strong oxidants, acids.

SECTION 8: Exposure controls/personal protection

Control parameters

Occupational Exposure limit values

TLV: 0.5 ppm as TWA.MAK: 1.9 mg/m3, 0.5 ppm; peak limitation category: II(2); skin absorption (H); sensitization of skin (SH); pregnancy risk group: C

Biological limit values

no data available

Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear safety goggles or eye protection in combination with breathing protection.

Skin protection

Protective gloves.

Respiratory protection

Use ventilation, local exhaust or breathing protection.

Thermal hazards

no data available

SECTION 9: Physical and chemical properties and safety characteristics

Physical state: Butyl mercaptan is a clear, colorless liquid with a strong skunk-like odor. Flash point in

range -18 to 43°F. Less dense than water and slightly soluble in water. Vapors heavier than

air.

Colorless liquid

Odour: Strong, obnoxious odor

Melting 58°C(lit.)

point/freezing

point:

Boiling point or 98°C(lit.)

initial boiling point and boiling range:

Flammability: Class IB Flammable Liquid: Fl.P. below 73°F and BP at or above 100°F.

Lower and upper

explosion

limit/flammability

limit:

Flash point: 12°C

Auto-ignition <225°C

temperature:

Decomposition

no data available

no data available

temperature:

pH: no data available

Kinematic

0.56 cP at 20 deg C

viscosity:

Solubility: 0.06 % (NIOSH, 2016)

Partition log Kow = 2.28

coefficient noctanol/water:

Vapour pressure: 83 mm Hg (37.7 °C)

Density and/or 0.842g/mLat 25°C(lit.)

relative density:

Relative vapour 3.1 (vs air)

density:

Particle no data available

characteristics:

SECTION 10: Stability and reactivity

Reactivity

Decomposes on heating. This produces toxic fumes of sulfur oxides (see ICSC 0074). Reacts with acids, bases and strong oxidants.

Chemical stability

no data available

Possibility of hazardous reactions

Dangerous by exposure to heat, flame, sparks or powerful oxidizers. The vapour is heavier than air and may travel along the ground; distant ignition possible. BUTYL MERCAPTAN is incompatible with acids, diazo and azo compounds, halocarbons, isocyanates, aldehydes, alkali metals, nitrides, hydrides, and other strong reducing agents. Reactions with these materials generate heat and in many cases hydrogen gas. May liberate hydrogen sulfide upon decomposition or reaction with an acid. Incompatible with strong oxidizing agents (USCG, 1999).

Conditions to avoid

no data available

Incompatible materials

Strong oxidizers (such as dry bleaches), acids.

Hazardous decomposition products

When heated to decomposition it emits /sulfur oxides/.

SECTION 11: Toxicological information

Acute toxicity

Oral: LD50 Mouse oral 3000 mg/kg

Inhalation: LC50 Mouse inhalation 2500 ppm/4 hr

Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

no data available

Reproductive toxicity

no data available

STOT-single exposure

The substance is irritating to the eyes, skin and respiratory tract. The substance may cause effects on the thyroid. Exposure far above the OEL could cause effects on the nervous system and lowering of consciousness.

STOT-repeated exposure

no data available

Aspiration hazard

A harmful contamination of the air can be reached very quickly on evaporation of this substance at 20°C.

SECTION 12: Ecological information

Toxicity

Toxicity to fish: LC50 Lepomis macrochirus 7.4 mg/L/24 hr /Conditions of bioassay not specified in source examined

Toxicity to daphnia and other aquatic invertebrates: no data available

Toxicity to algae: EC50 Scenedesmus subspicatus (Green algae; growth inhibition) 1068.3-5478.24 mg/L/96 hr; static

Toxicity to microorganisms: no data available

Persistence and degradability

AEROBIC: Little information is available in the literature on the biodegradation of 1-butyl mercaptan. Alcaligenes faecalis, a microorganism identified as a member of activated sludge flora, oxidizes 1-butyl mercaptan and other low molecular weight mercaptans(1).

Bioaccumulative potential

An estimated BCF of 11 was calculated for 1-butyl mercaptan(SRC), using a log Kow of 2.28(1) and a regression-derived equation(2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organisms is low.

Mobility in soil

The Koc of 1-butyl mercaptan is estimated as 410(SRC), using a log Kow of 2.28(1) and a regression-derived equation(2). According to a classification scheme(3), this estimated Koc value suggests that 1-butyl mercaptan is expected to have moderate mobility in soil.

Other adverse effects

SECTION 13: Disposal considerations

Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

SECTION 14: Transport information

UN Number

ADR/RID: UN2347 (For reference only, please check.) IMDG: UN2347 (For reference only, please check.) IATA: UN2347 (For reference only, please check.)

UN Proper Shipping Name

ADR/RID: BUTYL MERCAPTAN (For reference only, please check.)
IMDG: BUTYL MERCAPTAN (For reference only, please check.)
IATA: BUTYL MERCAPTAN (For reference only, please check.)

Transport hazard class(es)

ADR/RID: 3 (For reference only, please check.)
IMDG: 3 (For reference only, please check.)
IATA: 3 (For reference only, please check.)

Packing group, if applicable

ADR/RID: II (For reference only, please check.)

IMDG: II (For reference only, please check.) IATA: II (For reference only, please check.)

Environmental hazards

ADR/RID: No IMDG: No IATA: No

Special precautions for user

no data available

Transport in bulk according to IMO instruments

no data available

SECTION 15: Regulatory information

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

EC Inventory

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Listed.

New Zealand Inventory of Chemicals (NZIoC)

Listed.

(PICCS)

Listed.

Vietnam National Chemical Inventory

Listed.

IECSC)

Listed.

Korea Existing Chemicals List (KECL)

Listed.

SECTION 16: Other information

Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

 ${\it HSDB-Hazardous\ Substances\ Data\ Bank,\ website:\ https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm}$

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website:

http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

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