### Chemical Book India

# Chemical Safety Data Sheet MSDS / SDS

# 2,4,6-trichloroaniline SDS

Revision Date: 2024-04-25 Revision Number: 1

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# SECTION 1: Identification of the substance/mixture and of the company/undertaking

### Product identifier

Product name: 2,4,6-trichloroaniline

CAS: 634-93-5

# Relevant identified uses of the substance or mixture and uses advised against

Relevant identified For R&D use only. Not for medicinal, household or other use.

uses:

Uses advised none

against:

## Company Identification

Company: Chemicalbook.in

Address: 5 vasavi Layout Basaveswara Nilayam Pragathi Nagar Hyderabad, India -500090

Telephone: +91 9550333722

## **SECTION 2: Hazards identification**

### Classification of the substance or mixture

Acute toxicity - Category 4, Oral Skin sensitization, Category 1 Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 2

# GHS label elements, including precautionary statements

Pictogram(s)



Signal word Warning

### Hazard statement(s)

H302 Harmful if swallowed H317 May cause an allergic skin reaction H411 Toxic to aquatic life with long lasting effects

### Precautionary statement(s)

#### Prevention

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P272 Contaminated work clothing should not be allowed out of the workplace.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

P273 Avoid release to the environment.

### Response

P301+P317 IF SWALLOWED: Get medical help.

P330 Rinse mouth.

P302+P352 IF ON SKIN: Wash with plenty of water/...

P333+P317 If skin irritation or rash occurs: Get medical help.

P321 Specific treatment (see ... on this label).

P362+P364 Take off contaminated clothing and wash it before reuse.

P391 Collect spillage.

### Storage

none

### Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and

regulations, and product characteristics at time of disposal.

### Other hazards which do not result in classification

no data available

# **SECTION 3: Composition/information on ingredients**

### Substance

Chemical name: 2,4,6-trichloroaniline

Common names and

2,4,6-trichloroaniline

synonyms:

CAS number: 634-93-5 EC number: 211-219-8

Concentration: 100%

### **SECTION 4: First aid measures**

### Description of necessary first-aid measures

#### If inhaled

Move the victim into fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration and consult a doctor immediately. Do not use mouth to mouth resuscitation if the victim ingested or inhaled the chemical.

# Following skin contact

Take off contaminated clothing immediately. Wash off with soap and plenty of water. Consult a doctor.

# Following eye contact

Rinse with pure water for at least 15 minutes. Consult a doctor.

# Following ingestion

Rinse mouth with water. Do not induce vomiting. Never give anything by mouth to an unconscious person. Call a doctor or Poison Control Center immediately.

### Most important symptoms/effects, acute and delayed

SYMPTOMS: Symptoms of exposure to this compound may include irritation of the skin and eyes. ACUTE/CHRONIC HAZARDS: This compound may be harmful by inhalation, ingestion or skin absorption. It is an irritant of the skin and eyes. When heated to decomposition it emits toxic fumes of carbon monoxide, carbon dioxide, nitrogen oxides, and hydrogen chloride gas. (NTP, 1992)

### Indication of immediate medical attention and special treatment needed, if necessary

no data available

# **SECTION 5: Firefighting measures**

# Suitable extinguishing media

Fires involving this material can be controlled with a dry chemical, carbon dioxide or Halon extinguisher. A water spray may also be used. (NTP, 1992)

### Specific hazards arising from the chemical

Flash point data for this chemical are not available. It is probably combustible. (NTP, 1992)

# Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

### **SECTION 6: Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

# Environmental precautions

Prevent further spillage or leakage if it is safe to do so. Do not let the chemical enter drains. Discharge into the environment must be avoided.

# Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

# **SECTION 7: Handling and storage**

# Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### Conditions for safe storage, including any incompatibilities

Store the container tightly closed in a dry, cool and well-ventilated place. Store apart from foodstuff containers or incompatible materials

# **SECTION 8: Exposure controls/personal protection**

# Control parameters

### Occupational Exposure limit values

no data available

### Biological limit values

no data available

### Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

# Individual protection measures, such as personal protective equipment (PPE)

### Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

# Skin protection

Wear fire/flame resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

# Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

### Thermal hazards

no data available

# SECTION 9: Physical and chemical properties and safety characteristics

Physical state: Solid. Crystalline.

Colour: Pale White to reddish.

Odour: no data available

**Melting** 75 - 82 °C. Atm. press.:974.4 hPa.

point/freezing

point:

Boiling point or > 240 °C. Atm. press.:970.3 hPa.

initial boiling point and boiling range:

Flammability: no data available

Lower and upper

no data available

explosion

limit/flammability

limit:

Flash point: 127 °C. Atm. press.:972.4 hPa.

Auto-ignition temperature:

no data available

temperature:

Decomposition

no data available

temperature:

pH: 5.6.

Kinematic no data available

viscosity:

Solubility: less than 1 mg/mL at 70.7° F (NTP, 1992)

Partition log Pow = 3.64. Temperature: 25 °C. Remarks: PH details not available.

coefficient noctanol/water:

Vapour pressure: 0 mm Hg. Temperature: 25 °C.

Density and/or 0.525 g/cm3. Temperature:27 °C.

relative density:

Relative vapour no data available

density:

Particle no data available

characteristics:

# **SECTION 10: Stability and reactivity**

### Reactivity

This compound may be sensitive to exposure to light and air. Insoluble in water.

# Chemical stability

no data available

### Possibility of hazardous reactions

2,4,6-TRICHLOROANILINE is incompatible with acids, acid chlorides, acid anhydrides, chloroformates, and strong oxidizing agents. (NTP, 1992).

### Conditions to avoid

no data available

# Incompatible materials

no data available

# Hazardous decomposition products

When heated to decomposition it emits very toxic fumes of /hydrogen chloride/ and nitroxides.

# **SECTION 11: Toxicological information**

# Acute toxicity

Oral: LD50 - rat - 5 800 mg/kg bw. Remarks: No data available.

Inhalation: LCO - mouse - 1.5 mg/L air (analytical).

Dermal: LD50 - rabbit (male/female) - 2 462.038 mg/kg bw.

### Skin corrosion/irritation

no data available

# Serious eye damage/irritation

no data available

# Respiratory or skin sensitization

no data available

# Germ cell mutagenicity

no data available

# Carcinogenicity

no data available

# Reproductive toxicity

no data available

# STOT-single exposure

no data available

# STOT-repeated exposure

no data available

## Aspiration hazard

no data available

# **SECTION 12: Ecological information**

### **Toxicity**

Toxicity to fish: LC50 - Danio rerio (previous name: Brachydanio rerio) - > 8.51 - < 18.72 mg/L - 96 h.

Toxicity to daphnia and other aquatic invertebrates: EC50 - Daphnia magna - 2.3 mg/L - 48 h.

Toxicity to algae: EC50 - Chlorella vulgaris - 4.445 mg/L - 72 h.

Toxicity to microorganisms: EC50 - Photobacterium phosphoreum - 4.6 mg/L - 30 min.

### Persistence and degradability

AEROBIC: Using water samples containing bacteria acclimated to 2,4,6-trichloroaniline, a first-order rate decay of 0.206 day-1 was determined for the degradation of 2,4,6-trichloroaniline(1), with a half-life of 3.36 days(SRC). Negative inductive effects given by chlorosubstitutions in 2,4,6-positions operated strongly and prevented the enzymic conversion of 2,4,6-trichloroaniline using peroxidase enzymes(2). In a soil-bound form, residues of 2,4,6-trichloroaniline underwent aerobic mineralization more readily by soil micro flora than in a free state(3). Chemical binding of aniline residues to the soil organic matter effectively competes with the polymerization process by soil bacteria. Chemical binding to humic compounds acts greatly to retard mineralization(4).

### Bioaccumulative potential

A BCF of 3,630 was measured for 2,4,6-trichloroaniline(1) when exposed to fish for a 96 hr period. This suggests the potential for bioconcentration in aquatic organisms is very high, according to a classification scheme(2).

# Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc for 2,4,6-trichloroaniline can be estimated to be about 2,400(SRC). According to a classification scheme(2), this estimated Koc value suggests that 2,4,6-trichloroaniline is expected to have slight mobility in soil. Also, covalent binding to humic substances is an important process in the formation of soil-bound residues of chlorinated anilines. Additionally, other mechanisms of soil-binding, such as entrapping of residues into cavities

within the humus structure, or binding to mineral constituents, may also contribute to the phenomenon of bound residues of anilines in soil(3). Aromatic amines (such as various trichloro aniline isomers) have been observed to undergo rapid and reversible covalent bonding with humic materials in aqueous solution. The initial bonding reaction is followed by a slower and much less reversible reaction believed to represent the addition of the amine to quinoidal structures followed by oxidation of the product to give an amino-substituted quinone. These processes represent pathways by which aromatic amines may be converted to latent forms in the biosphere(4). There appears a rapid and spontaneous binding to soil organic matter and to purified humic and fulvic acid preparations. Therefore, a build up of 2,4,6-trichloroaniline in agricultural soils may occur(5).

### Other adverse effects

no data available

# **SECTION 13: Disposal considerations**

### Disposal methods

#### Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

# Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

### **UN Number**

ADR/RID: Not dangerous goods. (For reference only, please check.) IMDG: Not dangerous goods. (For reference only, please check.) IATA: Not dangerous goods. (For reference only, please check.)

# **UN Proper Shipping Name**

ADR/RID: Not dangerous goods. (For reference only, please check.) IMDG: Not dangerous goods. (For reference only, please check.) IATA: Not dangerous goods. (For reference only, please check.)

# Transport hazard class(es)

ADR/RID: Not dangerous goods. (For reference only, please check.) IMDG: Not dangerous goods. (For reference only, please check.) IATA: Not dangerous goods. (For reference only, please check.)

## Packing group, if applicable

ADR/RID: Not dangerous goods. (For reference only, please check.) IMDG: Not dangerous goods. (For reference only, please check.) IATA: Not dangerous goods. (For reference only, please check.)

### **Environmental hazards**

ADR/RID: Yes IMDG: Yes IATA: Yes

### Special precautions for user

no data available

# Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

# **EC Inventory**

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Listed.

New Zealand Inventory of Chemicals (NZIoC)

Not Listed.

(PICCS)

Not Listed.

Vietnam National Chemical Inventory

Not Listed.

IECSC)

Listed.

Korea Existing Chemicals List (KECL)

Listed.

### **SECTION 16: Other information**

# Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

http://www.echemportal.org/echemportal/index?pageID=0&request\_locale=en

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website:

http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

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