### Chemical Book India

# Chemical Safety Data Sheet MSDS / SDS

## 1,1-dichloroethylene SDS

Revision Date: 2024-04-25 Revision Number: 1

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# SECTION 1: Identification of the substance/mixture and of the company/undertaking

### Product identifier

Product name: 1,1-dichloroethylene

CAS: 75-35-4

# Relevant identified uses of the substance or mixture and uses advised against

Relevant identified For R&D use only. Not for medicinal, household or other use.

uses:

Uses advised none

against:

## Company Identification

Company: Chemicalbook.in

Address: 5 vasavi Layout Basaveswara Nilayam Pragathi Nagar Hyderabad, India -500090

Telephone: +91 9550333722

## **SECTION 2: Hazards identification**

### Classification of the substance or mixture

Flammable liquids, Category 1 Acute toxicity - Category 4, Inhalation

## Carcinogenicity, Category 2

### GHS label elements, including precautionary statements

Pictogram(s)







Signal word Danger

### Hazard statement(s)

H224 Extremely flammable liquid and vapour

H332 Harmful if inhaled

H351 Suspected of causing cancer

### Precautionary statement(s)

#### Prevention

P210 Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.

P233 Keep container tightly closed.

P240 Ground and bond container and receiving equipment.

P241 Use explosion-proof [electrical/ventilating/lighting/...] equipment.

P242 Use non-sparking tools.

P243 Take action to prevent static discharges.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P271 Use only outdoors or in a well-ventilated area.

P203 Obtain, read and follow all safety instructions before use.

### Response

P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse affected areas with water [or shower]. P370+P378 In case of fire: Use ... to extinguish.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P317 Get medical help.

P318 IF exposed or concerned, get medical advice.

### Storage

P403+P235 Store in a well-ventilated place. Keep cool.

P405 Store locked up.

# Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

#### Other hazards which do not result in classification

no data available

# **SECTION 3: Composition/information on ingredients**

#### Substance

Chemical name: 1,1-dichloroethylene

Common names and

1,1-dichloroethylene

synonyms:

CAS number: 75-35-4

EC number: 200-864-0

Concentration: 100%

# **SECTION 4: First aid measures**

## Description of necessary first-aid measures

#### If inhaled

Fresh air, rest. Refer immediately for medical attention.

## Following skin contact

Remove contaminated clothes. Rinse and then wash skin with water and soap.

### Following eye contact

Rinse with plenty of water (remove contact lenses if easily possible).

### Following ingestion

Rinse mouth. Do NOT induce vomiting. Refer for medical attention.

### Most important symptoms/effects, acute and delayed

Vapor can cause dizziness and drunkenness; high levels cause anesthesia. Liquid irritates eyes and skin. (USCG, 1999)

### Indication of immediate medical attention and special treatment needed, if necessary

Immediate first aid: Ensure that adequate decontamination has been carried out. If patient is not breathing, start artificial respiration, preferably with a demand-valve resuscitator, bag-valve-mask device, or pocket mask, as trained. Perform CPR as necessary. Immediately flush contaminated eyes with gently flowing water. Do not induce vomiting. If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain an open airway and prevent aspiration. Keep patient quiet and maintain normal body temperature. Obtain medical attention. Chlorinated fluorocarbons (CFCs) and related compounds

# **SECTION 5: Firefighting measures**

## Suitable extinguishing media

Use dry chemical, foam, carbon dioxide, or water spray. Use water spray to keep fire-exposed containers cool. Use flooding quantities of water. Fight fire from protected location or maximum possible distance. Vinylidene chloride, inhibited

### Specific hazards arising from the chemical

Special Hazards of Combustion Products: Toxic hydrogen chloride and phosgene are generated in fires. Behavior in Fire: May explode in fire due to polymerization. Vapor is heavier than air and may travel considerable distance to a source of ignition and flash back. (USCG, 1999)

### Special protective actions for fire-fighters

Use water spray, powder, foam, carbon dioxide. In case of fire: keep drums, etc., cool by spraying with water.

# **SECTION 6: Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

Evacuate danger area! Remove all ignition sources. Consult an expert! Personal protection: filter respirator for organic vapours of low boiling point adapted to the airborne concentration of the substance. Do NOT wash away into sewer. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

### **Environmental precautions**

Evacuate danger area! Remove all ignition sources. Consult an expert! Personal protection: filter respirator for organic vapours of low boiling point adapted to the airborne concentration of the substance. Do NOT wash away into sewer. Do NOT let this chemical enter the environment. Collect leaking and spilled liquid in sealable containers as far as possible. Absorb remaining liquid in sand or inert absorbent. Then store and dispose of according to local regulations.

### Methods and materials for containment and cleaning up

Land spill: Dig a pit, pond, lagoon, or holding area to contain liquid or solid material. /SRP: If time permits, pits, ponds, lagoons, soak holes, or holding areas should be sealed with an impermeable flexible membrane liner. / Dike surface flow using soil, sand bags, foamed polyurethane, or foamed concrete. Absorb bulk liquid with fly ash, cement powder, sawdust, or commercial sorbents. Apply appropriate foam to diminish vapor and fire hazard. Vinylidene chloride, inhibited

# **SECTION 7: Handling and storage**

### Precautions for safe handling

NO open flames, NO sparks and NO smoking. Closed system, ventilation, explosion-proof electrical equipment and lighting. Use non-sparking handtools. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

## Conditions for safe storage, including any incompatibilities

Store only if stabilized. Fireproof. Keep in the dark. Cool. Separated from incompatible materials. See Chemical Dangers. Store in an area without drain or sewer access. Provision to contain effluent from fire extinguishing. PROTECT AGAINST PHYSICAL DAWAGE. OUTSIDE OR DETACHED STORAGE IS PREFERABLE. INSIDE STORAGE SHOULD BE IN STD FLAWWABLE LIQUIDS STORAGE ROOM OR CABINET. SEPARATE FROM OXIDIZING MATERIALS.

# SECTION 8: Exposure controls/personal protection

### Control parameters

# Occupational Exposure limit values

TLV: 5 ppm as TWA; A4 (not classifiable as a human carcinogen). MAK: 8.0 mg/m3, 2 ppm; peak limitation category: II(2); carcinogen category: 3B; pregnancy risk group: C.EU-OEL: 8 mg/m3, 2 ppm as TWA; 20 mg/m3, 5 ppm as STEL

# Biological limit values

no data available

# Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

## Individual protection measures, such as personal protective equipment (PPE)

### Eye/face protection

Wear safety spectacles or eye protection in combination with breathing protection.

### Skin protection

Protective gloves.

# Respiratory protection

Use ventilation, local exhaust or breathing protection.

### Thermal hazards

no data available

# SECTION 9: Physical and chemical properties and safety characteristics

Physical state: Liquid.

Colour: Colorless.

Odour: Mild, sweet odor resembling that of chloroform

Melting -122.6 °C.

point/freezing

point:

Boiling point or 31.7 °C. Atm. press.:760 mm Hg.

initial boiling point and boiling range:

Flammability: Class IA Flammable Liquid: Fl.P. below 73°F and BP below 100°F.

Lower and upper

Lower flammable limit: 6.5% by volume; Upper flammable limit: 15.5% by volume

explosion

limit/flammability

limit:

Flash point: -28 °C.

Auto-ignition 570 °C. Remarks: Not measured/tested.

temperature:

**Decomposition** no data available

temperature:

pH: no data available

Kinematic dynamic viscosity (in mPa s) = 0.448. Temperature: -20.0 °C.; dynamic viscosity (in mPa s) =

viscosity: 0.394. Temperature:0.0°C.;dynamic viscosity (in mPa s) = 0.33. Temperature:20°C.

**Solubility:** 5 to 10 mg/mL at 70° F (NTP, 1992)

Partition log Pow = 2.12. Remarks: No temperature or pH provided.

coefficient noctanol/water:

Vapour pressure: 102.7 hPa. Temperature: 20 °C.; 286.6 hPa. Temperature: 0 °C.; 665 hPa. Temperature: 20

°C.

Density and/or relative density:

1.21 g/cm3. Temperature:20 °C.

Relative vapour

3.46 (vs air)

density:

Particle no data available

characteristics:

# **SECTION 10: Stability and reactivity**

### Reactivity

NIOSH considers vinylidene chloride to be a potential occupational carcinogen.

The substance can readily form explosive peroxides. The substance readily polymerizes due to heating or under the influence of oxygen, sunlight, copper or aluminium. This generates fire or explosion hazard. May explode on heating or on contact with flames. Decomposes on burning. This produces toxic and corrosive fumes of hydrogen chloride and phosgene. Reacts violently with oxidants.

### Chemical stability

no data available

# Possibility of hazardous reactions

Flammable liquidThe vapour is heavier than air and may travel along the ground; distant ignition possible. Vapours are uninhibited and may polymerize, causing blockage of vents. Peroxidizable monomer, such as VINYLIDENE CHLORIDE, may initiate exothermic polymerization of the bulk material [Bretherick 1979. p. 160, 187]. Mixing vinylidene chloride in equal molar portions in a closed container with any of the following substances caused the temperature and pressure to increase: chlorosulfonic acid, nitric acid, or oleum [NFPA 1991]. Its reaction products with ozone are particularly dangerous [Dow Chemical, 1968]. This may extend to other powerful oxidants, as various peroxides are produced.

#### Conditions to avoid

no data available

### Incompatible materials

At ambient temp, perchloryl fluoride is unreactive with 1,1-dichloroethylene, but reaction is explosive at 100-300 deg C, or if the mixture is ignited.

## Hazardous decomposition products

When not stabilized, decomp in air into chlorine, hydrogen chloride, phosgene, & formaldehyde & white polymeric powder

## **SECTION 11: Toxicological information**

Acute toxicity

Oral: LD50 Mouse oral approx 200 mg/kg

Inhalation: LC50 Rat (nonfasted) inhalation 10000-15000 ppm/4 hr

Dermal: no data available

### Skin corrosion/irritation

no data available

### Serious eye damage/irritation

no data available

### Respiratory or skin sensitization

no data available

### Germ cell mutagenicity

no data available

### Carcinogenicity

Evaluation: There is inadequate evidence in humans for the carcinogenicity of vinylidene chloride. There is limited evidence in experimental animals for the carcinogenicity of vinylidene chloride. Overall evaluation: Vinylidene chloride is not classifiable as to its carcinogenicity to humans (Group 3).

## Reproductive toxicity

No studies were located regarding developmental or reproductive effects in humans. Birth defects were noted in the offspring of pregnant rats and mice that had been exposed to vinylidene chloride in air. In this study, maternal toxicity was observed at developmentally toxic concentrations.

## STOT-single exposure

The substance is mildly irritating to the eyes and upper respiratory tract. Exposure far above the OEL could cause lowering of consciousness.

### STOT-repeated exposure

The substance may have effects on the kidneys and liver. This may result in liver function impairment and kidney impairment. Tumours have been detected in experimental animals but may not be relevant to humans.

# Aspiration hazard

A harmful contamination of the air can be reached very quickly on evaporation of this substance at 20°C.

# **SECTION 12: Ecological information**

### **Toxicity**

Toxicity to fish: LC10 - Pimephales promelas - 92.6 mg/L - 3 d.

Toxicity to daphnia and other aquatic invertebrates: EC50 - Daphnia magna - 37 mg/L - 48 h.

Toxicity to algae: EC50 - Chlamydomonas reinhardtii - 9.12 mg/L - 72 h.

Toxicity to microorganisms: EC10 - Pseudomonas putida - > 2 000 mg/L - 17 h.

## Persistence and degradability

AEROBIC: 1,1-Dichloroethylene, present at 9.7 mg/L, reached 0% of its theoretical BOD in 4 weeks using an activated sludge inoculum at 2 mg/L and the Closed Bottle Test(1). In another study, 45-78% of the chemical was lost in 7 days when incubated with a wastewater inoculum; however, a sizeable fraction of the loss was due to volatilization(2). 97% of 1,1-dichloroethylene was reported to be removed in a municipal wastewater plant(3). 1,1-Dichloroethylene at a concn of 160 ug/L was degraded using continuous mixed, batch-fed reactors at a 20-day operating solids retention time(4). 1,1-Dichloroethylene had a biodegradation half-life of 1.25 yrs in ground water from a former manufacturing facility in NJ(5).

### Bioaccumulative potential

BCFs of 2.5 to 6.4 and <13 were measured for 1,1-dichloroethylene at concentrations of 0.5 and 0.05 mg/L, respectively(1). According to a classification scheme(2), these BCF values suggest bioconcentration in aquatic organisms is low(SRC).

## Mobility in soil

The Koc for 1,1-dichloroethylene has been measured as 64(1) and 65(2). According to a classification scheme(3), this Koc value suggests that 1,1-dichloroethylene is expected to have high mobility in soil. A Koc of 293 was measured in sewage solids(4).

### Other adverse effects

no data available

## **SECTION 13: Disposal considerations**

# Disposal methods

### **Product**

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

### Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

# **SECTION 14: Transport information**

### **UN Number**

ADR/RID: UN1303 (For reference only, please check.) IMDG: UN1303 (For reference only, please check.) IATA: UN1303 (For reference only, please check.)

## **UN Proper Shipping Name**

ADR/RID: VINYLIDENE CHLORIDE, STABILIZED (For reference only, please check.)
IMDG: VINYLIDENE CHLORIDE, STABILIZED (For reference only, please check.)
IATA: VINYLIDENE CHLORIDE, STABILIZED (For reference only, please check.)

# Transport hazard class(es)

ADR/RID: 3 (For reference only, please check.) IMDG: 3 (For reference only, please check.) IATA: 3 (For reference only, please check.)

# Packing group, if applicable

ADR/RID: I (For reference only, please check.)
IMDG: I (For reference only, please check.)
IATA: I (For reference only, please check.)

### Environmental hazards

ADR/RID: No IMDG: No IATA: No

# Special precautions for user

no data available

## Transport in bulk according to IMO instruments

no data available

# **SECTION 15: Regulatory information**

Safety, health and environmental regulations specific for the product in question

European Inventory of Existing Commercial Chemical Substances (EINECS)

Listed.

**EC Inventory** 

Listed.

United States Toxic Substances Control Act (TSCA) Inventory

Listed.

China Catalog of Hazardous chemicals 2015

Listed.

New Zealand Inventory of Chemicals (NZIoC)

Listed.

(PICCS)

Listed.

Vietnam National Chemical Inventory

Listed.

IECSC)

Listed.

Korea Existing Chemicals List (KECL)

Listed.

## **SECTION 16: Other information**

### Abbreviations and acronyms

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

#### References

IPCS - The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home

HSDB - Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm

IARC - International Agency for Research on Cancer, website: http://www.iarc.fr/

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website:

 $http://www.echemportal.org/echemportal/index?pageID=0 \\ \texttt{Strequest\_locale=en}$ 

CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple

ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website:

http://www.phmsa.dot.gov/hazmat/library/erg

Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis-stoffdatenbank/index-2.jsp

ECHA - European Chemicals Agency, website: https://echa.europa.eu/

#### Other Information

Depending on the degree of exposure, periodic medical examination is suggested. An added stabilizer or inhibitor can influence the

toxicological properties of this substance; consult an expert. The odour warning when the exposure limit value is exceeded is insufficient. Do NOT use in the vicinity of a fire or a hot surface, or during welding.

Disclaimer: The above information is believed to be correct but does not purport to be all inclusive and shall be used only as a guide. The information in this document is based on the present state of our knowledge and is applicable to the product with regard to appropriate safety precautions. It does not represent any guarantee of the properties of the product. We as supplier shall not be held liable for any